

Research Article

Design and Synthesis of Novel Vicinal Diaryl 1,2,4-triazole Derivatives with Potential Antiepileptic Activity

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Abstract

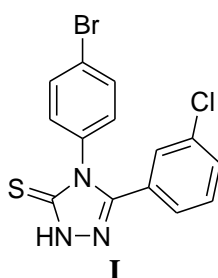
A vicinal diaryl 1,2,4-triazole derivative that has the structure feature of antiepileptic drugs has been prepared and tested using maximal electroshock test (MES). The target compound gives 0/10 versus 9/10 protection ratio for phenytoin.

Key words: Antiepileptic drugs, vicinal diaryl, 1,2,4-triazole.

Introduction

Epilepsy is a serious neurological disease in the brain. As reported by WHO, more than 50 million were affected by epilepsy worldwide. Epilepsy is characterized by recurrent seizures due to abnormal excessive discharge of cerebral neurons.^[1] Epileptic seizures appear as involuntary movement, sensation, or thoughts. The disease is caused by various factors as cerebral organic lesion, head trauma, neurological disease, metabolic disorders, alcohol abuse, drug overdose or toxicities. Although antiepileptic mechanism is complex, there is some common chemical features between antiepileptic drugs as Hydrophobic domain (A), Electron donor atom (D), and Hydrogen acceptor /donor unit (HAD).^[1]

Hydrophobic domain is either substituted or unsubstituted aromatic ring, electron rich atom and hydrogen acceptor /donor unit and distal hydrophobic domain but not essential. Anticonvulsant activity is one of the most important biological activities of vicinal diphenyltriazoles. Luszczki et al., synthesized and evaluated anticonvulsant activity of 4-(4-bromophenyl)-6-(3-chlorophenyl)-2,4-dihydro-3H-1,2,4-triazole-5-thione (I), which significantly potentiated the anticonvulsant effect of phenobarbital, carbamazepine and sodium valproate. The mechanism of action had an apparent resemblance to sodium channel antagonists, especially to lamotrigine.



The targeted compound 4 has these features; carboxylic group acts as (HAD), vicinal diaryl rings act as hydrophobic group have two proximal, and both amide nitrogen atom and N7 of triazole ring can act as electron rich atom.

According to the above found results, the present work involves design, synthesis and evaluation of novel vicinal diaryl 1,2,4-triazole derivative for its anticonvulsant activity in maximal electroshock (MES)

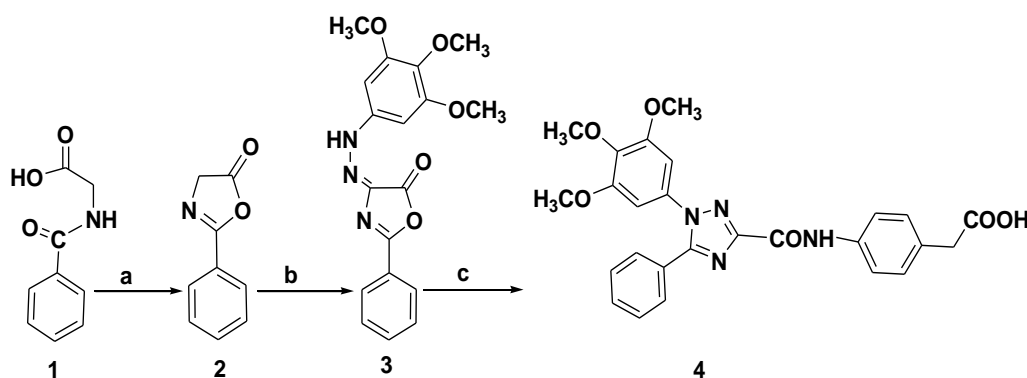
Result and discussion

Chemistry

Key starting compound, γ -(benzamido) acetic acid **1** was prepared in high yield (80%) by the reaction of glycine with benzoyl chloride in 10% NaOH. Heating of compound **1** with acetic anhydride afforded the corresponding lactone. The methylene group of lactone possesses nucleophilic character and the degree of nucleophilicity depends on the electron withdrawing effect of the adjacent carbonyl group.

The synthesis of the key intermediate compound **2** was carried out using Kuskov like reaction through coupling of the diazonium salt of 3,4,5-trimethoxyaniline with the active methylene of lactone in presence of sodium acetate. The diazonium

salt acts as an electrophile which was attacked by the carbanion derived from the active methylene group in lactone. This reaction is usually carried out in cold aqueous solution buffered with sodium acetate. The reaction yields an unstable azo compound, which spontaneously tautomerizes to hydrazone derivative compound **3** in 70% yields. The target compound **4** was obtained in a good yield by Sawdey rearrangement.^[7] The reaction involves attack of a nucleophile, γ -(ϵ -aminophenyl) acetic acid with opening of the oxazoline ring to form acyclic carboxamide derivatives, which then cyclized via loss of water molecule to afford 1,2,4-triazole- γ -carboxamide derivative **4** as shown in scheme 1.



Reagents and conditions: a) Ac_2O , 70°C , 2 min. b) ϵ -chloroaniline, HCl, NaOAc, 70°C , 2 hr. c) γ -(ϵ -aminophenyl)acetic acid, AcOH, NaOAc, reflux 2 hr

Scheme 1. Synthesis of compound **4**

Biological evaluation (*in vivo* antiepileptic activity)

The resulted anticonvulsant activity of compounds **4** was performed using phenytoin and sodium valproate as reference. Compound **4** showed the same protection activities at 0.5 hr and 2 hr (50%

protection rate versus 50% for phenytoin) (Table 1). It was observed that compound **4** showed no activity against seizures for subcutaneous pentylenetetrazole (scPTZ) pretreated mice while it had a weak activity against seizures for strychnine induced seizures.

Table (1): Anticonvulsant activities of compound 4

Compound	Anticonvulsant Activity					Neurotoxicity ^d	
	MES ^a		scPTZ ^b	Strychnine ^c		0/1 (h)	1/1 (h)
	0/1 (hr)	1/1 (hr)		Onset of convulsion	Time of death		
Control	0/10	0/10	0%	1.1 ± 0.12	3.11 ± 0.20	0/10	0/10
Compound 4	0/10	0/10	0%	1.22 ± 0.10	3.11 ± 0.17	0/10	0/10
Phenytoin sodium	9/10	8/10	N.D.	N.D.	N.D.	3/10	2/10
Sodium valproate	N.D. ^e	N.D.	80	0.48 ± 0.12	1.12 ± 0.28	1/10	0/10

^aMaximal electroshock test (number of animals protected / number of animals tested)

^bSubcutaneous pentylenetetrazole test (% protection)

^cIntraperitoneal strychnine test (onset of convulsions and time to death)

^dRotarod test (number of animals exhibiting toxicity / number of animals tested)

^eNot Determined

Experimental

Chemistry

Materials and methods

Melting points were determined on Stuart electro-thermal melting point apparatus and are uncorrected. IR spectra were recorded on Nicolet iS^o FT-IR spectrometer at Minia University. ¹H NMR spectra were carried out using Bruker apparatus 400 MHz spectrometer, using TMS as internal reference. High resolution mass spectra (HRMS) were obtained on a Thermo Scientific Q ExactiveTM Orbitrap mass spectrometer at The University of British Columbia Canada. Reactions were routinely monitored by thin-layer chromatography (TLC) using Merck 9380 pre-coated aluminum plate silica gel (Kieselgel 60) 0.25 x 20 cm plates with a layer thickness of 0.2 mm, and spots were visualized by exposure to UV-lamp at $\lambda = 254$ nm.

Materials: Chemicals and solvents used in the preparation of the target compounds are of commercial grade, and purchased from Aldrich, Merck, Fluka, Cambrian chemicals, and El-Nasr pharmaceutical Chemicals Companies.

Synthesis of 2-phenyl-4H-oxazol-5-one (2)

Hippuric acid 1 (2.33 g, 0.013 mol) in acetic anhydride (5.0 ml) was heated until a clear solution was obtained then this solution was cooled to room temperature.

Synthesis of 4-(3,4,5-trimethoxyphenyl)-hydrazono-2-phenyl-2-oxazol-5-one (3)

Acetic anhydride (5 mL) was added to hippuric acid 1 (0.013 mol, 2.33 g). The mixture was heated until a clear yellow solution was obtained. The solution was cooled to room temperature to form solution A. A mixture of 3,4,5-trimethoxyaniline (0.013 mol, 2.38 g), and HCl (0.1 M, 2 mL) was kept in an ice-salt bath. A solution of sodium nitrite (0.013 mol, 0.89 g) in water (1 mL) was added in a drop wise manner, and the mixture was stirred for 10 min. Anhydrous sodium acetate (0.013 mol, 1.0 g) was added to the mixture to form solution B. Solution A was added to solution B in a drop wise manner and stirring was continued at 0-10°C for 2 hr. The formed precipitate was filtered off, washed with distilled water and dried to give compound 3 as red precipitate (yield 50%); IR (KBr, cm⁻¹): 1784 (C=O), 1630 (C=N), 1577 (C=C), 1230 (C-O-C)

Synthesis of 2-(4-(1-(3,4,5-trimethoxyphenyl)-2-phenyl-1H-1,2,4-triazole-3-carboxamido)phenyl) acetic acid (4)

A mixture of compound 2 (0.1 mol, 2.66 g), 2-(ϵ -aminophenyl) acetic acid (0.12 mol), and anhydrous sodium acetate (0.70 g) were dissolved in glacial acetic acid (20 mL). The reaction mixture was stirred under reflux for 2 hr, and was left to cool, and was poured to ice water. The precipitate formed was filtered off, washed with water, and recrystallized by reaction with saturated solution of sodium bicarbonate and warmed gently the resulted solution was filtered and other organic impurities with organic solvent, the aqueous solution was acidified with dil. HCl then filtered under vacuum and collected. Yellowish white crystals (3.8g, 78%); IR ν (cm⁻¹): 3668(OH), 3302 (NH), 1717 (C=O), 1678 (C=O), 1590 (C=N); ¹H-NMR (ϵ 0.0 MHz, DMSO-*d*₆) δ (ppm): 3.60 (t, 2H, HOOC-CH₂), 3.71 (s, 2H, 2 OCH₂), 3.89 (s, 2H, 2 OCH₂), 6.61 (s, 2H, 2 Ar-H), 7.42-7.70 (m, 1.0H, 1.0 Ar-H), 7.93 (d, 2H, *J* = 8 Hz, H_{7,8} of benzoic acid ring), 8.00 (d, 2H, *J* = 8 Hz, H_{5,6} of benzoic acid ring), 9.08 (s, 1H, NH); ¹³C-NMR (100 MHz, DMSO-*d*₆) δ (ppm): 40.16, 56.20, 64.00, 103.04, 120.03, 126.70, 127.87, 128.70, 129.07, 130.00, 130.71, 132.49, 136.61, 138.70, 153.02, 154.80, 156.21, 156.62, 170.77 HRMS: m/z calculated for C₁₇H₁₄N₂O₂ [M-H]⁺: 287.10231, found: 287.10266

Biological evaluation

Maximal electroshock (MES) model

Thirty male albino mice weighting 20-30 g were divided into three equal groups each of ten. The first group was used as a control group administered dimethyl sulfoxide (DMSO) only as a vehicle intraperitoneally. The second group was administered phenytoin sodium. The third was administered test compound. Compound 2 and phenytoin sodium were dissolved in DMSO and injected intraperitoneally to the animal at dose 100 mg/kg body weight one hour before MES test. Seizures were induced by means of 60 Hz current of 60 mA delivered through ear electrodes [Hugo Basile-Italy]. The stimulus duration was 0.2 s and pulse width was 0.4. The criterion to indicate the convulsion response was the hind limb tonic extension (HLTE). The presence or absence of HLTE was noted. Animal in which extensor response was abolished were taken as protected mice.^[12-13]

Evaluation of *in vivo* (antiepileptic activity Subcutaneous pentylenetetrazole (scPTZ) model)

Thirty male albino mice weighting 20-30 g were divided into three equal groups each of five and acclimatized to their environment for at least one week before the experiment. The first group was used as a control group administered DMSO only as a vehicle intraperitoneally. The second group was administered sodium valproate. The third group was administered test compounds.

The test compound and sodium valproate were dissolved in DMSO and injected intraperitoneally to the animal at dose 100 mg/kg body weight. One hour later, mice were injected with PTZ 40 mg/kg body weight in scruff of the neck.

The dose of PTZ was selected by preliminary screening as lower dose failed to induce typical seizures while higher doses only increased the mortality. Animals devoid of generalized convulsions were considered to be protected and the results were represented as protection percent.^[14]

Intraperitoneally strychnine HCl induced convulsions

Thirty male albino mice weighting 20-30 g were divided into three equal groups each of ten. The first group was used as a control group administered DMSO only as a vehicle intraperitoneally. The second group was administered sodium valproate.

The third group was administered test compound. Compound 2 and sodium valproate were dissolved in DMSO and injected intraperitoneally to the animal at dose 100 mg/kg body weight. One hour later, mice were injected with strychnine HCl at dose 2 mg/kg body weight intraperitoneally. The onset of convulsions and time of death was calculated in comparison with control group.^[15]

Neurotoxicity (Rotarod test)

Minimal motor impairment was measured in mice by rotarod test. Thirty male albino mice weighting 20-30 g were trained to stay

on accelerating rotarod that rotates at 10 r.p.m., the rod diameter was 2.2 cm. and trained animals were divided into four equal groups each of ten. The first group was used as a control group administered DMSO only as a vehicle intraperitoneally. The second group was administered phenytoin sodium. The third group was administered sodium valproate. The fourth group was administered test compound. Compound 4, phenytoin sodium and sodium valproate were dissolved in DMSO and injected intraperitoneally to the animal at dose 100 mg/kg body pweight. Neurotoxicity was indicated by the inability of the animals to maintain equilibrium on the rod for at least 1 minute in each of three trials.^[1]

Conclusion

The similarities of the selected compound to phenytoin not only on the structural features required for anticonvulsant activity, but also it resembles phenytoin on its mechanism of anticonvulsant activity. Compound 4 have MES induced seizures inhibition less than phenytoin. It also has minimal activity against strychnine and scPTZ induced convulsions. Pentylene-tetrazole and isoniazid have been reported to produce seizures by inhibiting γ -aminobutyric acid (GABA) neurotransmission.^[3] Compound 4 failed to protect animals from seizure induced by Strychnine at the dose of 300 mg/kg. It is known that Strychnine directly antagonizes the inhibitory spinal reflexes of glycine,^[1] so the results suggest that compound 4 could neither influence glycine system nor gabergic neuro-transmission.

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